organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 180 K Mean σ (C–C) = 0.004 Å Disorder in main residue R factor = 0.060 wR factor = 0.168 Data-to-parameter ratio = 14.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2,6-Bis[2,4-bis(heptyloxy)phenyl]pyridine

The title 2,6-disubstituted pyridine, $C_{41}H_{61}NO_4$, with a crystallographic twofold axis, has an arrangement of molecules well organized to undergo multiple cyclometallation reactions.

Received 16 December 2004 Accepted 13 January 2005 Online 22 January 2005

Comment

2,6-Disubstituted pyridines are ideally set up to undergo multiple cyclometallations, reactions of considerable interest to us (Cave *et al.*, 1999, 2000). In addition to their ability to undergo multiple cyclometallations, such compounds have also been shown to be activated by other reagents (Cave *et al.*, 1998).



The title molecule, (I) (Fig. 1), has crystallographic twofold symmetry and the aliphatic chains are each in an extended form. Within the unit cell, the molecules are aligned in an antiparallel fashion (Figs. 2 and 3).

Experimental

1-Chloro-2,4-bis(heptyloxy)benzene (15.0 g, 48.0 mmol) was added dropwise to a stirred solution of magnesium (1.22 g, 50.0 mmol) and methyl iodide (0.30 g, 2.11 mmol) in tetrahydrofuran (THF, 25 ml) under an inert atmosphere. The resulting Grignard reagent was transferred via a cannula to a stirred solution of 2,6-dichloropyridine (2.96 g, 20.0 mmol) and tetrakis(triphenylphosphine)palladium (0.45 g, 0.50 mmol) in THF (25 ml). The reaction mixture was then heated under reflux (24 h) under an inert atmosphere. Excess Grignard was destroyed with water (20 ml) and hydrochloric acid (5 ml, 2 M). The neutralized (aqueous NaOH) reaction mixture was extracted with diethyl ether (2 \times 200 ml) and dried (saturated NaCl and magnesium sulfate). The solvent was removed under vacuum and the product crystallized out as white needle-like crystals (yield: 12.0 g, 19.0 mmol, 95%). ¹H NMR (CDCl₃, 250.13 MHz): δ 7.96 (2H, d, ³J = 8.5 Hz), 7.75 (2H, d, ${}^{3}J$ = 7.3 Hz), 7.60 (1H, t, ${}^{3}J$ = 7.3 Hz), 6.59 (2H, dd, ${}^{3}J = 8.5$, ${}^{4}J = 2.1$ Hz), 6.53 (2H, d, ${}^{4}J = 2.1$ Hz), 3.99 (8H, t, ${}^{3}J =$ 6.7 Hz), 1.79 (8H, m), 1.48 (24H, m), 0.89 (12H, m). ¹³C NMR (CDCl₃, 250.13 MHz): δ 160.5, 157.7, 154.7, 134.8, 132.1, 131.5, 121.9, 105.6, 100.0, 68.2, 31.4, 25.7, 22.5. Elemental analysis found (expected): C 77.8 (77.9), H 9.6 (9.7), N 2.4% (2.2%).

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Figure 1

View of the title molecule, showing the atomic numbering. Displacement ellipsoids are drawn at the 50% probability level. Both disorder components are shown in all Figures.

 $D_x = 1.117 \text{ Mg m}^{-3}$

Cell parameters from 2201

Mo $K\alpha$ radiation

reflections

T = 180(2) K

 $R_{\rm int}=0.045$

 $\theta_{\rm max} = 25.0^{\circ}$

 $\begin{array}{l} h=-13\rightarrow 13\\ k=-8\rightarrow 7 \end{array}$

 $l = -18 \rightarrow 28$

Block, colourless

 $0.45 \times 0.25 \times 0.20 \text{ mm}$

1753 reflections with $I > 2\sigma(I)$

 $\begin{array}{l} \theta = 3 - 15^{\circ} \\ \mu = 0.07 \ \mathrm{mm}^{-1} \end{array}$

Crystal data

 $\begin{array}{l} C_{41}H_{61}NO_4 \\ M_r = 631.91 \\ Monoclinic, P2/n \\ a = 11.2502 \ (10) \ {\rm \AA} \\ b = 6.9682 \ (6) \ {\rm \AA} \\ c = 24.014 \ (2) \ {\rm \AA} \\ \beta = 93.963 \ (2)^{\circ} \\ V = 1878.0 \ (3) \ {\rm \AA}^3 \\ Z = 2 \end{array}$

Data collection

Bruker SMART diffractometer ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{\min} = 0.896, T_{\max} = 0.989$ 8996 measured reflections 3318 independent reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.060$	H-atom parameters constrained $w = 1/[\sigma^2(F^2) + (0.078P)^2]$
$wR(F^2) = 0.168$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.04	$(\Delta/\sigma)_{\rm max} = 0.043$
3318 reflections	$\Delta \rho_{\rm max} = 0.21 \text{ e} \text{ Å}^{-3}$
229 parameters	$\Delta \rho_{\rm min} = -0.22 \text{ e } \text{\AA}^{-3}$

H atoms were added at calculated positions (C-H = 0.95–0.99 Å) and refined using a riding model (including free rotation about C-C bonds for methyl groups), with $U_{iso}(H) = 1.2$ (or 1.5 for methyl H atoms) times $U_{eq}(C)$. The terminal section of one of the heptane chains was found to be disordered between two positions [relative occupancies 0.43 (1):0.57 (1)]. This disorder is believed to be responsible for various anomalies in the displacement parameters of the atoms in this region of the molecule.

Data collection: *SMART* (Siemens, 1994); cell refinement: *SAINT* (Siemens, 1995); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.



Figure 2

Packing diagram of (I), viewed down the a axis. H atoms have been omitted.





Packing diagram of (I), viewed down the b axis. H atoms have been omitted.

The EPSRC and Siemens generously supported the purchase of the SMART diffractometer.

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